Tetrahedron Letters No.18, pp. 1255-1259, 1965. Pergamon Press Ltd. Printed in Great Britain.

CHEMICAL TRANSFORMATION OF (+)-HIBAENE INTO (-)-KAURENE

A.H. Kapadi and Sukh Dev National Chemical Laboratory, Poona (India)

(Received 4 March 1965)

Special interest attaches to the conversion of hibaene (I) into kaurene (II), as both of these are biogenetically linked through the precursor III, which in its stereochemically specified state⁺ is the biogenetic progenitor for the various tetracarbocyclic diterpenes according to Wenkert scheme¹. Though, at the time of the

*Communication No. 767 , National Chemical Lab., Poona.



hibaene kaurene atisine and trachylobane⁷ types



phyllocladene and other so far unknown⁸ types. original proposal no diterpene of type I was known with certainty², several members of this class have been described^{3,4,5,6} in recent years. We now report the chemical



transformation of (+)-hibaene (I)⁶ into (-)-kaurene (II)⁹.

(+)-Hibaene epoxide⁶ on treatment with BP_3 -Et₂⁰ in benzene at 0° gave in 85% yield an unsaturated alcohol, $C_{20}H_{32}^{0}$ (m.p. 134-135°, [\propto]_D -30.3°) with the following structural features: three quaternary methyl groups (PMR^{*}: 3H sharp singlets at 50, 52 and 62 cps); one vinylic methyl (PMR: 3H doublet centred at 103 cps, J = 2 cps); one olefinic proton (PMR: 1H signal centred at 295 cps; IR: 1650, 865 cm⁻¹; positive tetranitromethane test); one secondary hydroxyl (PMR: 1H signal, essentially a singlet, at 240 cps; IR: 3400, 1052, 1044 cm⁻¹); these characteristics⁺ are in accord with the structure IV, anticipated

All PMR spectra were measured in $\sim 5-10\%$ soln. in CCl₄ on a Varian A-60 Spectrometer; the signals are recorded in cps from tetramethylsilane (internal standard) as zero.

⁺Cf. e.g. the Me signals of isokaurene:48,51,61 and 100 cps.

mechanistically. The above alcohol on oxidation with $Cr0_3$ -pyridine furnished the unsaturated ketone (V, m.p. 113-114°, [<]_D +16.0°; $y^{c=0}$ 1750 cm⁻¹; $\lambda_{max}^{sthanol}$ 273, C 31, also a shoulder at 228 with C 810) which, as required by structure V, showed the shielding of the C_{10} -Me by the carbonyl in its PMR spectrum (all the three quaternary methyls showing a composite 9H signal at 52 cps). The ketone on Wolff-Kishner reduction¹⁰ gave in 62% yield a



mixture of dihydrokaurene^{*} (m.p. 84-85[°], $[<]_D$ -30[°]) and kaurene (II)^{*} (m.p. 50-50.5[°], $[<]_D$ -78.5[°]), readily

^{*} The reduction of an olefinic linkage during Wolff-Kishner reduction has been reported; for references see: R.Soman, Sukh Dev, R.Misra and R.C.Pandey, <u>Tetrahedron Letters</u> 3767 (1964).

^{*}Apparently isomerisation of isokaurene to kaurene occurs with facility under these conditions.

separated over AgNO3-silica gel¹¹ and identified by standard procedures.

The above rearrangement (Hibaene \rightarrow isokaurene type) appears to be general. Thus (-)-monogynyl acetate¹² (19-acetory-I) could be readily converted into 19-acetoxy-IV (m.p. 151-152°, [\propto]_D²⁷ = -31.7°) in an overall yield of 66% by this route.

In connection with the above work it should be mentioned that (+)-hibaene on being refluxed for 2.5 hr in a 1:1 mixture of dioxane-formic acid, was recovered essentially unchanged; only traces of some formate were produces. On treatment of (+)-hibaene with hydrogen chloride in C_6H_6 -CHCl₃ at 0° the corresponding hydrochloride (m.p. 73-74°; [<]_D -29°. PMR: four quaternary methyl signals at 49, 51, 56 and 58.5 cps; -CHCl- signal, as a multiplet centred at 265 cps) was obtained.

Acknowledgement. The authors wish to record their indebtedness to Prof. G. Ourisson for a sample of isokaurene and Prof. B.E. Cross for a specimen of (-)-kaurene.

REFERENCES

- 1 E. Wenkert, <u>Chem. and Ind.</u> 282 (1955); W.B. Whalley, <u>Tetrahedron</u> 18, 43 (1962).
- 2 Mirene has since been shown to be a mixture of kaurene and phyllocladene: C.Djerassi, P.Quitt, E. Mosettig, R.C.Cambie, P.S.Butledge and L.H.Briggs, <u>J.Amer.Chem.</u> <u>Soc. 83</u>, 3720 (1961).
- 3 P.R.Jefferies, R.S.Rosich, D.E.White and M.C.Woods, <u>Austral. J. Chem. 15</u>, 521 (1962).
- 4 W.H. Baarschers, D.H.S.Horn and L.F.Johnson, <u>J.Chem.</u> Soc. 4046 (1962).
- 5 Y. Kitahara and A. Yoshikoshi, Tet.Letters 1771(1964).
- 6 A.H.Kapadi and Sukh Dev, <u>Tet. Letters</u> 1171,2751(1964); R.D.H.Murray and R. McCrindle, <u>Chem.and Ind</u>.500(1964).
- 7 G. Hugel, L. Lods, J.M.Mellor, D.W.Theobald and G. Ourisson, <u>Bull. Soc. Chim. Fr.</u> 1974 (1963).
- 8 E. Wenkert, P.W.Jeffs and J.R. Mahajan, <u>J. Amer. Chem.</u> <u>Soc.</u> <u>86</u>, 2218 (1964).
- L.H. Briggs, B.F.Cain, R.C.Cambie, B.R.Davis, P.S. Rutledge and J.K. Wilmshurst, <u>J. Chem. Soc.</u> 1345(1963);
 S. Masamune, <u>J. Amer. Chem. Soc</u>. <u>86</u>, 289 (1964).
- 10 D.H.R. Barton, D.A.J. Ives and B.R.Thomas, <u>J.Chem.Soc.</u> 2056 (1955).
- 11 N.P. Damodaran and Sukh Dev, <u>Tet. Letters</u> 1941 (1963).
- 12 A.H. Kapadi and Sukh Dev, Tet. Letters 1171 (1964).